

## PYRROLIZIDINE ALKALOIDS FROM *SENECIO CACALIASTER*\*

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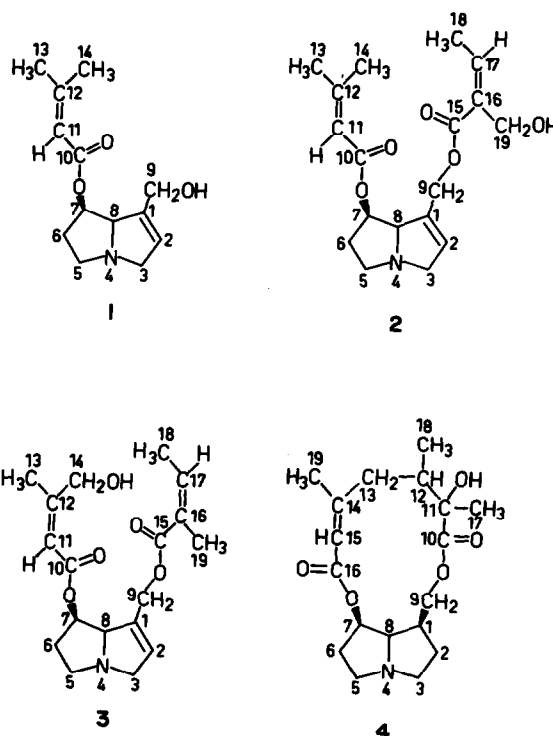
**Abstract**—Four pyrrolizidine alkaloids have been isolated from *Senecio cacaliaster* and their structures analysed by spectroscopic methods (IR, mass, <sup>1</sup>H, <sup>13</sup>CNMR). One of them is new and the name sencalenine (3) is proposed. Alkaloids O<sup>7</sup>-senecioldretronecine (1) and 7-seneciold-9-sarracinyldretronecine (2) have recently been identified elsewhere. The fourth is bulgarsenine (4) which was isolated from a *Senecio* species before.

### INTRODUCTION

*Senecio cacaliaster* is widely distributed in Europe. On account of the fact that this plant possesses no ligules, Linné adjoined it to the genus *Cacalia* and named it *S. cacaliaster*. However, in spite of the missing ligules this plant does resemble *S. nemorensis*, especially *S. nemorensis* subsp. *fuchsii* which may be implied from its outward shape. Both of them belong to the Betulo adenostyletea [1]. Ecologically *S. cacaliaster* is closely related to *S. nemorensis* subsp. *fuchsii*, from which the pyrrolizidine alkaloids (PA) fuchsisenecionine and senecionine could be isolated [2, 3]. *S. lamottei* is a hybrid of these two species. Therefore, a common precursor of *S. cacaliaster* and *S. nemorensis* has to be considered. We investigated *S. cacaliaster* with regard to its alkaloids in order to prove this relationship. We detected four major alkaloids by TLC, which we then separated by droplet counter current chromatography (DCCC).

### RESULTS AND DISCUSSION

Methanolic extraction of plant material was followed by purification of the residue as already described [3]. From the resulting residue four PAs have been isolated by DCCC. The mass spectra gave the formulae C<sub>13</sub>H<sub>19</sub>NO<sub>3</sub> for 1, C<sub>18</sub>H<sub>25</sub>NO<sub>5</sub> for 2 and 3 and C<sub>18</sub>H<sub>27</sub>NO<sub>5</sub> for 4. The fragmentation of the latter alkaloid is identical to that of bulgarsenine [4]. A typical fragmentation of 1–3 between *m/z* 136 and *m/z* 80 is characteristic for retronecine or its isomeric form. In addition, the fragmentations show that PAs 2 and 3 are isomeric diesters, whereas PA 1 must be a retronecine monoester. The mass spectrum also gives information about the esterification of PA 1 at C-7 with a 5-C-linked acid. The alkaloids 2 and 3, having similar 5-C-linked acids, are esterified at both alcoholic positions, whereas in 2, 3 and 4 two such bonds were found. Further information about the structures is given by NMR analysis (Tables 1 and 2).



The NMR data for 1 are reported for the first time; in the case of alkaloid 2 our data differ from those described recently [5] by interchange of the <sup>13</sup>C-values for carbons 1/12/16 and 3/9 and 13/14. Likewise, there is some difference in the <sup>1</sup>H NMR data especially in the values for the C-5 H<sub>2</sub>. All values were verified by decoupling experiments and by evaluation of coupled and noise-decoupled spectra. The value for C-6 H<sub>2</sub> is higher by 2 ppm for all three alkaloids. This also demonstrates that retronecine is the necic base [6, 7]. The NMR data indicate the structure of an O<sup>7</sup>-seneciold-acid-retronecine-ester for PA 1. This is clear from the <sup>1</sup>H peaks at δ 1.89 and 2.16 for two methyls and the corresponding

\*Dedicated to Professor Maximilian Steiner on the occasion of his 80th birthday.

Table 1.  $^1\text{H}$  NMR data of 1, 2 and 3 ( $\text{CDCl}_3$ ; TMS)

	1	2	3
$\text{C}_{13}\text{-H}_3$	1.89, d, 3H $J = 1$	1.87, d, 3H $J = 1$	2.04, d, 3H
$\text{C}_{14}\text{-H}_3$	2.16, d, 3H $J = 1$	2.05, d, 3H $J = 1$	—
$\text{C}_{14}\text{-H}_2$	—	—	3.62, m, 2H
$\text{C}_{14}\text{-OH}$	—	—	4.73, m, OH
$\text{C}_6\text{-H}_2$	2.06, m, 2H	2.67, q, 2H $J = 9$	2.33, m, 2H
$\text{C}_5\text{-H}_A$	2.71, m, 1H	3.31, m, 1H	2.29, m, 1H
$\text{C}_3\text{-H}_A$	3.28, m, 1H	3.49, m, 1H	3.1, m, 1H
$\text{C}_5\text{-H}_B$	3.38, m, 1H	3.67, m, 1H	3.6, m, 1H
$\text{C}_3\text{-H}_B$	3.89, m, 1H	3.93, m, 1H	3.75, m, 1H
$\text{C}_{18}\text{-H}_3$	—	2.02, d, 3H $J = 7.5$	1.97, 2q, 3H $J = 7; 1.5$
$\text{C}_9\text{-H}_2$	4.1, d, 2H	4.78, m, 2H	4.71, m, 2H
$\text{C}_{19}\text{-H}_3$	—	—	1.87, q, 3H $J = 1.5$
$\text{C}_{19}\text{-H}_2$	—	4.2, s, 2H	—
$\text{C}_{19}\text{-OH}$	—	4.39, m, OH	—
$\text{C}_8\text{-H}$	4.7, m, 1H	4.51, m, 1H	4.44, m, 1H
$\text{C}_9\text{-OH}$	4.57, m, OH	—	—
$\text{C}_7\text{-H}$	5.45, m, 1H	5.4, m, 1H	5.62, m, 1H
$\text{C}_{11}\text{-H}$	5.6, m, 1H	5.6, m, 1H	5.97, m, 1H
$\text{C}_2\text{-H}$	5.68, m, 1H	5.78, m, 1H	5.78, m, 1H
$\text{C}_{17}\text{-H}$	—	6.38, q, 1H $J = 7.5$	6.16, 2q, 1H $J = 7$

 $\delta$  values in ppm;  $J$  in Hz.Table 2.  $^{13}\text{C}$  NMR data of 1, 2 and 3 ( $\text{CDCl}_3$ ; TMS)

Carbon No.	1	2	3
18	—	15.8	15.9
13	20.46	20.3	20.6
14	27.58	27.5	66.8
6	34.31	34.4	34.2
5	53.47	53.7	54.0
9	59.43	60.8	60.5
3	61.8	62.3	62.1
19	—	64.7	16.0
7	72.44	72.9	71.8
8	76.32	75.7	76.4
11	115.0	115.8	123.8
2	121.8	127.0	126.8
1	138.79	131.8	140.2
12	138.75	133.9	133.6
17	—	141.0	128.1
16	—	158.0	160.8
15	—	165.6	164.9
10	168.0	166.6	167.18

\* $\delta$  values in ppm.

olefinic proton at 5.6. The structure of 1 is also verified from the corresponding  $^{13}\text{C}$  data. The NMR spectra of PA 2 show partially the same peaks as the NMR data of 1; thus, 1 is found as a partial structure in 2. Furthermore, in 2 the esterified acid at C-9 must be a hydroxyangelic acid.

This is confirmed by a quartet at  $\delta$  2.02 with a coupling constant of 7.5 Hz for C-18  $\text{H}_3$  and the olefinic proton at 6.38 with the same coupling constant. The hydroxyl group must be located at position C-19 on account of the value  $\delta$  4.2 for the methylene group. This structure is also established by the following  $^{13}\text{C}$  data:  $\delta$  64.7 for C-19 and 15.8 for C-18. The data for compound 3 show that it is an isomer of 2. Instead of peaks for the methyl group at C-14 those of a hydroxymethyl group at  $\delta$  4.73 ( $^1\text{H}$ ) and 66.8 ( $^{13}\text{C}$ ) are found. In comparison with 2, peaks for an angelic acid are present in 3 for the acid at C-9: C-18  $\text{H}_3$ :  $\delta$  1.97 ( $J = 7; 1.5 \text{ Hz}$ ) =  $^1\text{H}$  and 15.9 =  $^{13}\text{C}$  NMR; C-19  $\text{H}_3$ :  $\delta$  1.87 ( $J = 1.5 \text{ Hz}$ ) =  $^1\text{H}$  and 16.0 =  $^{13}\text{C}$  NMR. The esterification of 2 and 3 as shown in the formulae is proved by mass spectrometry. The fragmentation pattern of 2 shows an intense ion at  $m/z$  220. This ion is formed after ester cleavage at C-9 building a necine, esterified at C-7 with a 3-methyl-2-butenic acid. Furthermore, ion  $m/z$  219 shows a McLafferty rearrangement and a loss of 100 mu ( $m/z$  119). The ion  $m/z$  237 also proves the structure 2. It is formed after rearrangement of the ester grouping at C-9 and the loss of  $\text{C}_5\text{H}_6\text{O}_2$  indicating the hydroxymethyl group position at C-1. In the fragmentation pattern of 3 the ion  $m/z$  219 is formed after McLafferty rearrangement and the loss of 116 mu. Only alkaloids which possess an open diester form show such rearrangements at C-7. Therefore, in 3 there must be an acid of MW 116 esterified at C-7. Furthermore, an ester cleavage is found at C-9 forming the ion  $m/z$  236. This esterification at C-7 and C-9 in 3 is established by the fact that the ion  $m/z$  220 is missing. For the fourth alkaloid (= bulgarsenine) we found similar data as reported earlier [4, 8]. For the new alkaloid 3 we propose the name sencialenine.

## EXPERIMENTAL

*S. cacaliaster* (Lam.) was collected in alpine Austria (Highlands of Grossglockner). The dried and pulverized drug was extracted and purified as mentioned earlier [3]. The resulting yellow residue was separated by DCCC using toluene-MeOH- $\text{H}_2\text{O}-\text{CHCl}_3$  in the ascending mode yielding the four PAs [9]. Content of PAs (GC; dry wt): 1, 0.017%; 2, 0.005%; 3, 0.005%; 4, 0.21%. GC conditions: glass column: 1.8 m, 2 mm; 4% SE 30 on Varaport 30; prog: 80–240° in four steps at 9°/min:  $2 \times 2^\circ/\text{min}$ ;  $1 \times 3^\circ/\text{min}$ ;  $1 \times 4^\circ/\text{min}$ . Inj.: 200°. Det.: 260°; carrier gas ( $\text{N}_2$ ): 40 ml/min.  $R_s$  values (min): 1, 1.53; 2, 21.8; 3, 17.6; 4, 20.15.

*O<sup>7</sup>-Senecierylretronecine* (1). Oily;  $[\alpha]_D^{20} = -2$  (EtOH). IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3385 (OH), 3010 (C=C), 1730 (satd ester), 1705 (unsatd ester), 1655 (C=C). CIMS ( $\text{CHCl}_3$ ), 70 eV,  $m/z$  (% rel. int.): 237.137  $[\text{M}]^+$  (10.29), 219  $[\text{M}-\text{H}_2\text{O}]^+$  (1.47), 154  $[\text{M}-\text{C}_5\text{H}_7\text{O}]^+$  (11.05), 137  $[\text{M}-\text{C}_5\text{H}_8\text{O}_2]^+$  (38.64), 136  $[\text{M}-\text{C}_5\text{H}_7\text{O}]^+$  (30.76), 124  $[\text{M}-\text{CH}_2\text{O}]^+$  (34.31), 111  $[\text{M}-\text{C}_2\text{H}_2]^+$  (47.09), 106  $[\text{M}-\text{CH}_2-\text{OH}]^+$  (48.94), 94  $[\text{M}-\text{OH}]^+$  (24.77), 80  $[\text{M}-\text{CH}_2-\text{OH}]^+$  (100.0).

*7-Senecieryl-9-sarracinyltretronecine* (2). Oily;  $[\alpha]_D^{20} = +4$  (EtOH). IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3390 (OH), 3010 (C=C), 1720 (satd ester), 1705 (unsatd ester), 1645 (C=C). CIMS ( $\text{CHCl}_3$ ), 70 eV,  $m/z$  (% rel. int.): 335.173  $[\text{M}]^+$  (9.06), 237  $[\text{M}-\text{C}_5\text{H}_8\text{O}_2]^+$  (42.32), 221  $[\text{M}-\text{C}_5\text{H}_8\text{O}_3]^+$  (7.08), 220  $[\text{M}-\text{C}_5\text{H}_7\text{O}_3]^+$  (40.85), 219  $[\text{M}-\text{C}_5\text{H}_8\text{O}_3]^+$  (8.79), 138  $[\text{M}-\text{C}_5\text{H}_7\text{O}]^+$  (12.81), 137  $[\text{M}-\text{C}_5\text{H}_7\text{O}]^+$  (11.96), 136  $[\text{M}-\text{C}_5\text{H}_7\text{O}]^+$  (100.0), 121  $[\text{M}-\text{C}_5\text{H}_8\text{O}_2]^+$  (17.59), 120  $[\text{M}-\text{C}_5\text{H}_8\text{O}_2]^+$  (37.08), 119  $[\text{M}-\text{C}_5\text{H}_8\text{O}_2]^+$  (28.93), 106  $[\text{M}-\text{CH}_3\text{O}]^+$  (8.67), 95  $[\text{M}-\text{C}_2\text{H}_2]^+$  (7.83), 94  $[\text{M}-\text{C}_2\text{H}_2]^+$  (46.57), 93  $[\text{M}-26]^+$  (67.31).

*Senecalene* (3). Oily;  $[\alpha]_D^{20} = -8$  (EtOH). IR  $\nu_{\max}^{\text{KBr}} \text{ cm}^{-1}$ : 3380 (OH), 3060 (C=C), 1730 (satd ester), 1710 (unsatd ester), 1660 (C=C). CIMS ( $\text{CHCl}_3$ ) 70 eV,  $m/z$  (rel. int.): 335.172  $[\text{M}]^+$  (13.50), 252  $[\text{M} - \text{C}_5\text{H}_7\text{O}_2]^+$  (22.04), 236  $[\text{M} - \text{C}_5\text{H}_7\text{O}_2]^+$  (28.46), 235  $[\text{M} - \text{C}_5\text{H}_8\text{O}_2]^+$  (15.16), 219  $[\text{M} - \text{C}_5\text{H}_8\text{O}_3]^+$  (12.32), 154  $[\text{M} - \text{C}_5\text{H}_6\text{O}_2]^+$  (20.18), 137  $[\text{M} - \text{C}_5\text{H}_6\text{O}]^+$  (11.88), 136  $[\text{M} - \text{C}_5\text{H}_7\text{O}_2]^+$  (100.0), 122  $[\text{M} - \text{C}_5\text{H}_7\text{O}_3]^+$  (28.10), 120  $[\text{M} - \text{C}_5\text{H}_7\text{O}_2]^+$  (52.06), 119  $[\text{M} - \text{C}_5\text{H}_8\text{O}_3]^+$  (31.13), 106  $[\text{M} - \text{CH}_2 - \text{OH}]^+$  (9.40), 94  $[\text{M} - \text{C}_2\text{H}_2]^+$  (60.88), 93  $[\text{M} - \text{C}_2\text{H}_2]^+$  (79.55), 80  $[\text{M} - \text{C}_3\text{H}_6]^+$  (18.27).

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